Listing of Claims

This listing of claims will replace all prior versions and listings of claims in the Application.

- 1. (CURRENTLY AMENDED) A device for sensing NO_x compounds comprising:
 - a calix[4]arene compound, wherein the calix[4]arene compound forms eapable of forming a stable and reversible complex with at least one NO⁺ cation, wherein a detectable charge-transfer reaction occurs between the at least one NO⁺ cation and the calix[4]arene detectable as a color change, wherein NO⁺ is derived from an oxide of nitrogen in a form other than nitric oxide.
- 2. (ORIGINAL) The device of claim 1, wherein the detection is selected from the group consisting of visualization, measurement of electrochemical changes, and measurement of spectroscopic changes.
- 3. (ORIGINAL) The device of claim 1, wherein the complex undergoes dissociation.
 - 4. (ORIGINAL) The device of claim 3, wherein the complex is decolorized.
- 5. (ORIGINAL) The device of claim 1, wherein the calix[4] arene compound is alternatively a cone calix[4] arene, a 1, 3-alternate calixarene or a combination thereof.
- 6. (ORIGINAL) The device of claim 1, wherein the calix[4] arene compound is optionally immobilized, in solution, attached to a ligand, attached to a solid support, or any combination thereof.
- 7. (CURRENTLY AMENDED) The device of claim 1, wherein the chargetransfer reaction is detected in the presence of mixtures selected from the group consisting of H₂O, O₂, HCl, SO_X, NH₃, NO, their derivatives and NO_x compounds are optionally a gas, liquid, solution, mixtures of gases, or a combinations thereof.
- 8. (ORIGINAL) The device of claim 1, wherein the complex is a storage device for the NO⁺ cation.
- 9. (ORIGINAL) The device of claim 1, wherein the complex is capable of transferring the NO⁺ cation to a substrate.

- 10. (ORIGINAL) The device of claim 1, wherein the complex is stabilized by one or more Lewis acids.
- 11. (CURRENTLY AMENDED) A device for purifying chemical compounds containing NO_x comprising:

a calix[4]arene compound, wherein the calix[4]arene compound <u>reversibly</u> complexes a NO⁺ cation from the chemical compound and is capable of transferring <u>purifies the chemical compound free of</u> the NO⁺ cation produced from the NO_x and wherein NO⁺ is derived from an oxide of nitrogen in a form other than nitric oxide.

- 12. (ORIGINAL) The device of claim 11, wherein the calix[4] arene compound is optionally immobilized, in solution, attached to a ligand, on a solid interface, attached to a solid support, or a combination thereof.
- 13. (ORIGINAL) The device of claim 11, wherein the complex is a storage device for the NO⁺ cation.
- 14. (ORIGINAL) The device of claim 13, wherein the complex is chemically stable for at least several weeks.
- 15. (CURRENTLY AMENDED) A method of purifying chemical compounds comprising:

exposing a calix[4] arene compound to a mixture of chemical species containing at least one NOx compound;

allowing the calix[4] arene compound to interact with the mixture, wherein the calix[4] arene compound forms a stable and reversible complex with an an NO⁺ from the NOx compound complex and wherein NO⁺ is derived from an oxide of nitrogen in a form other than nitric oxide, wherein a detectable charge-transfer reaction occurs between the NO⁺ cation and the calix[4] arene for identification of the NO⁺.

- 16. (CURRENTLY AMENDED) A molecular container comprising:

 a calix[4]arene compound enriched with electron donating groups; and
 at least one NO⁺ cation, wherein NO⁺ is derived from an oxide of nitrogen in a
 form other than nitric oxide, wherein the NO⁺ cation is attracted to an electron donating
 group, forms a stable and reversible complex with the calix[4]arene compound and
 wherein the NO+ cation is now made available for transfer to a second container that
 accepts the NO⁺ cation.
- 17. (ORIGINAL) The molecular container of claim 16, wherein the calix[4] arene compound complexes the NO⁺ cation and is capable of storing it.

- 18. (ORIGINAL) The molecular container of claim 16, wherein the calix[4] arene compound complexes the NO⁺ cation and is capable of transferring it to another substrate.
 - 19. (CURRENTLY AMENDED) An optical switch comprising:

a <u>reversible</u> and <u>stable</u> calix[4]arene-nitrosonium complex <u>formed</u> by noncovalent forces <u>between a calix[4]arene compound and free nitrosonium</u> in which the nitrosonium <u>switches</u> is <u>eapable of changing</u> between a free and <u>a</u> complexed state wherein the switching <u>is</u> <u>ean be</u> detected optically, wherein the nitrosonium is derived from an oxide of nitrogen in a form other than nitric oxide.

20. (CURRENTLY AMENDED) An optical switch comprising:

a means for <u>reversibly</u> complexing <u>in a stable form</u> a nitrosonium cation by noncovalent forces, wherein the nitrosonium is derived from an oxide of nitrogen in a form other than nitric oxide; and

a means for detecting the presence of the complexed nitrosonium cation.

- 21. (NEW) The optical switch of claim 19, wherein the calix[4] arene compound is immobilized on a solid support.
- 22. (NEW) The molecular container of claim 16, wherein the second container is selected from the group consisting of calix[4]arene compound, solid support, and solid interface.
 - 23. (NEW) An improved nitrosating agent comprising:

a compound that reversibly entraps one or more nitrosonium cations in a deep cavity by noncovalent forces and releases the one or more nitrosonium cations under reactive conditions, wherein the reactive conditions are selected from the group consisting of a change in temperature, solvent polarity, and cavity shape, wherein the compound is a a calix[4]arene compound selected from the group consisting of a cone and a 1, 3-alternate conformation specifically synthesized with a deep cavity.